

Modern NIR spectrometer technology for real-time state-of-charge estimation of heat storage materials

Evgeny Legotin^{1,2,*}, Gayaneh Issayan³, Bernhard Zettl³, Markus Brandstetter¹, Christian Rankl¹

¹ RECENDT – Research Center for Non-Destructive Testing GmbH, Linz, Austria

² TU Wien, Institute of Chemical, Environmental and Bioscience Engineering, Vienna, Austria

³ University of Applied Sciences Upper Austria, Energy Research Group ASIC, Wels, Austria

* Corresponding author: evgeny.legotin@recendt.at

SORPTION HEAT STORAGE

- Well-suited to store solar and geothermal energy, both for seasonal and daily demands
- Shows high energy densities, robustness, low heat losses, and good cycling stability as compared to sensible and latent heat storages
- Water is a commonly used sorbate.
- Important characteristics are adsorption capacity, cyclability, state-of-health (SOH), and state-of-charge (SOC). A suitable technology to control these parameters in real time is required.
- Only a few concepts of a suitable real-time SOC sensor exist.



Studied sorption heat storage materials: a) A-type zeolite; b) A-type zeolite, binder-free; c) NaY-type zeolite, binder-free; d) microporous silica gel; e) salt-hydrate composite, 20 wt% salt; f) salt-hydrate composite, 40 wt% salt

MOEMS-BASED NIR SPECTROMETER TECHNOLOGY

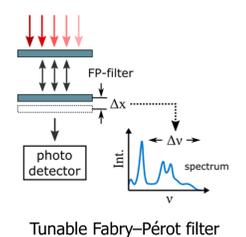


Benefits:

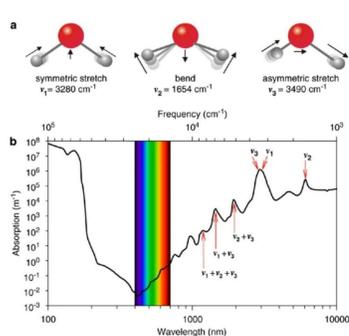
- miniature
- low-cost
- robust (therefore, low maintenance costs)
- flexible spectral range and step

Drawbacks:

- low spectral resolution
- narrow spectral range



CONCEPT



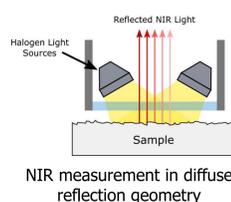
The absorption band at ca. 5150 cm^{-1} ($\lambda \approx 1940\text{ nm}$), representing a combination of bending (ν_2) and asymmetric stretching (ν_3) vibrations of water molecules, is particularly suitable for quantification of adsorbed water in solids.

◀ Fundamental vibrations of the H_2O molecule (a) and the absorption spectrum of pure water (b).

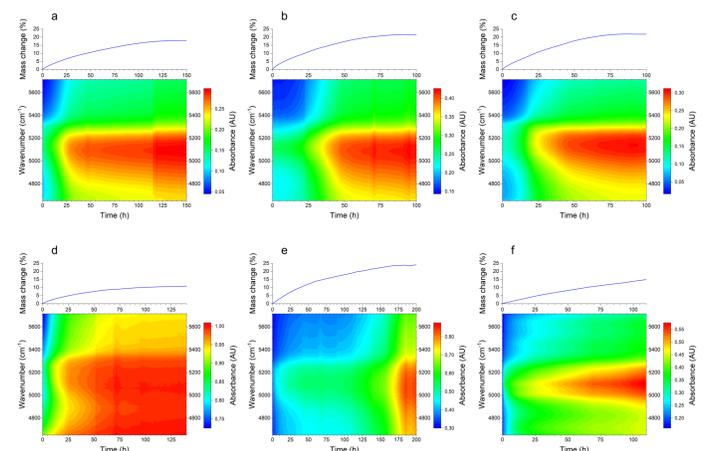
Stomp, M. *et al.* ISME J. 1 (2007) 271

EXPERIMENTAL METHOD

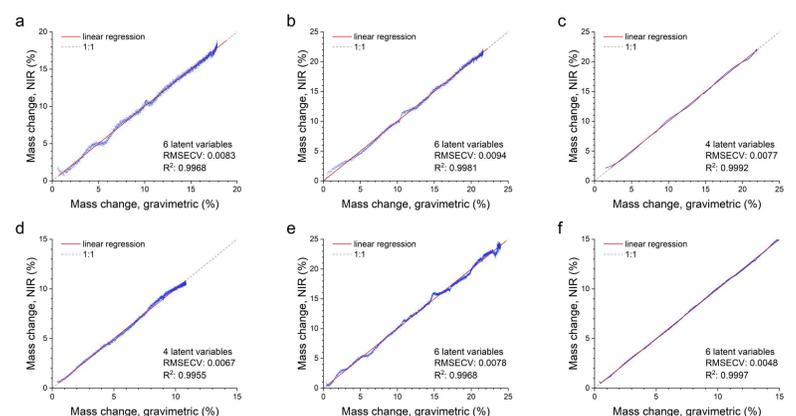
1. Charging (= desorption) in a drying chamber (zeolites: 3 h at $250\text{ }^\circ\text{C}$; silica gel: 3 h at $150\text{ }^\circ\text{C}$; salt-hydrate composites: 2 h at $100\text{ }^\circ\text{C}$ + 2 h at $200\text{ }^\circ\text{C}$)
2. Water vapor adsorption measurement with simultaneous mass change and NIR spectra collection: The spectra were collected in the spectral range of 1750–2150 nm using a NIR Fabry-Pérot microspectrometer with integrated light sources. The mass changes were recorded using an analytical balance ($\pm 0.0001\text{ g}$ precision).
3. Data preprocessing: smoothing, interpolation
4. PLS modelling with absorbances at each wavelength as the independent variables and the relative mass change (in %) as the dependent variable



RESULTS



NIR spectra and respective gravimetric curves for the studied materials: a) A-type zeolite; b) A-type zeolite, binder-free; c) NaY-type zeolite, binder-free; d) microporous silica gel; e) salt-hydrate composite, 20 wt% salt; f) salt-hydrate composite, 40 wt% salt



Predicted vs. reference plots for the studied materials: a) A-type zeolite; b) A-type zeolite, binder-free; c) NaY-type zeolite, binder-free; d) microporous silica gel; e) salt-hydrate composite, 20 wt% salt; f) salt-hydrate composite, 40 wt% salt

CONCLUSION

Based on water adsorption measurements, we developed PLS models showing high accuracy of water content prediction. Novel MOEMS-based NIR spectrometers can provide an adequately precise, compact, and low-cost solution for SOC control of various sorption heat storage systems. Our research findings are useful to assess heat storage capacities, improve balancing local heat supplies and demands, and therefore, increase the resilience of future heat distribution networks.

ACKNOWLEDGEMENT

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